

# Development of an advanced CFD model for ash deposit and aerosol formation in biomass fired boilers

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## ABSTRACT

A CFD based deposit formation model for biomass fired boilers has been developed. The model considers the condensation of ash vapours, deposition of coarse salt-rich and silica-rich fly ash particles, brittle and ductile erosion of the deposit layer by non-sticky particles, aerosol formation and deposition under consideration of a single particle size class. Furthermore, the influence of the growing deposit layer on heat transfer (heat conduction, radiative emissivity of the surfaces) in the furnace and the boiler is taken into account.

The model has been evaluated concerning its applicability as process analysis and design tool for biomass combustion plants. For this purpose, test runs with a small-scale, fixed bed pellet furnace (70 kW<sub>th</sub> nominal boiler capacity) and a medium-scale grate furnace with a 10 MW<sub>th</sub> thermal oil boiler (fuels: fibre board, sawdust) were performed. The modelling results were validated against deposit formation and dust emission measurements (mass and chemical composition) performed at different locations of the plants.

The numerical results show the interaction between the involved chemical reactions within the turbulent reactive flow, condensation, fine particle formation as well as deposition of coarse fly ash particles. While the description of the ash vapours regarding condensation and fine particle formation already delivers quantitatively correct results, the accuracy of the stickiness model of coarse fly ash particles has to be improved.

**Keywords:** biomass combustion, ash deposition, CFD simulation

## 1. INTRODUCTION AND OBJECTIVES

Nowadays, Computational Fluid Dynamics (CFD) has been well established as an engineering tool to optimise biomass furnaces and boilers regarding flow and gas phase combustion.

Since agricultural residues, olive residues or waste wood constitute an attractive alternative to common biomass fuels boiler manufacturers and operators have to find strategies to take into account the risk of ash related problems like slagging in furnaces, fouling as well as corrosion in boilers in order to be competitive. 3D CFD simulations can be applied for the purpose of a spatially resolved simulation and analysis of deposit and aerosol formation in dependence of relevant influencing parameters.

Therefore, at BIOENERGY 2020+ a detailed CFD model for ash deposit formation in biomass fired boilers has been developed. It considers the condensation of ash vapours, the deposition of coarse salt-rich and silica-rich fly ash particles, brittle and ductile erosion of the deposit layer by non-sticky particles, aerosol formation and deposition under consideration of a single particle size class as well as the influence of the growing deposit layer on heat transfer (heat conduction and radiative emissivity of the surfaces) in the boiler.

In order to evaluate the reliability and accuracy of the model, test runs for a 70 kW pellet fixed bed furnace and a 10 MW biomass grate furnace with fibre board and sawdust as fuels have been performed. The simulations were compared with measurements of deposited mass and chemical composition of the deposits in different plant zones as well as with dust emission measurements.

## 2. MODEL SETUP

### OVERVIEW

The deposit formation model has been implemented in the CFD Code FLUENT 12 using user defined functions (UDF). The calculation of solid biomass combustion on the grate and the release of flue gas components from the fuel bed is performed using an empirical packed bed combustion model based on experimental data [11] in order to provide the boundary conditions for the subsequent simulation of the turbulent reactive flow. The calculation of gas phase combustion is based on the Realizable  $k-\epsilon$  model for turbulence, the Discrete Ordinate Model for radiation and the Eddy Dissipation model in combination with a global methane 3-step reaction mechanism including the flue gas components  $\text{CH}_4$ ,  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{H}_2$ ,  $\text{H}_2\text{O}$  and  $\text{O}_2$  for gas phase combustion. A detailed description of the CFD combustion model for biomass grate furnaces can be found in [11].

The deposit build-up is simulated within a post-processing step based on the flow and gas phase combustion simulation. The empirical packed bed combustion model was extended in order to calculate the release of ash vapours and the entrainment of coarse fly ash particles from the grate to provide the boundary conditions for the CFD simulation of ash deposit and aerosol formation. Subsequently, CFD simulations of ash vapour condensation on the boiler walls, aerosol formation and deposition on the boiler walls are performed. Furthermore, the calculation of the deposition of coarse fly ash particles on boiler walls is based on two different stickiness approaches - a viscosity approach for silica-rich particles and a melting approach for salt-rich particles.

The condensation of the ash vapours as well as the deposition of coarse fly ash particles and fine particles contribute to the time-dependent build-up of the deposit layer. The growth of the deposit layer influences the heat transfer due to changes of the thermal resistance and emissivity of the layer as a function of its chemical composition, temperature, structure and thickness. Furthermore, erosion of deposits by coarse fly ash particles is considered.

In the following, a short description of the most relevant sub-models for ash deposit formation is given:

### WALL CONDENSATION AND FINE PARTICLE FORMATION

During solid fuel combustion, ash forming vapours are released from the fuel bed into the gas phase. The mass flux and the composition of the ash components depend on the temperature in the fuel bed as well as the composition of the fuel. Due to the cooling of the flue gas or the contact with cooled walls, the vapour pressure of the components decreases. If the vapour pressure becomes lower than the saturation pressure of the ash vapour, the aggregate state changes to liquid or solid phases. Consequently, fine particles are formed in the gas phase, which form deposits on walls or are emitted. Furthermore, when the vapour pressure on the wall is lower than in the boundary layer, condensation occurs.

Ash forming vapours are involved in a complex reaction system. Here, with exception of the kinetically limited formation of sulphates, it is assumed that reactions between ash forming vapours are fast in comparison to local residence time of the flue gas. In this case, the chemical composition can be determined by thermodynamic equilibrium calculations. The accuracy of the predicted compositions depends on the quality of the thermodynamic data of the ash components, which are reliable for most alkali salts [6]. In case of the formation of sulphates, a kinetic approach was applied. According to Christensen [4] the formation of  $\text{SO}_3$  from precursors ( $\text{SO}_2$ ) is limited rather than the final formation of sulphates. In this model, the global reaction rate according to Christensen based on an Arrhenius approach, was implemented in combination with the Eddy Dissipation Model. Furthermore, in order to reduce the mathematical effort to model the transport of the ash vapours, the deposit formation model considers the transport of the elements (C, S, Cl, K, Na) instead of ash forming species as these elements are the most relevant ones regarding aerosol formation in wood pellet boilers [3]. To calculate condensation fluxes to the walls or fine particle formation, ash vapour pressures of species are determined by thermodynamic equilibrium calculations in combination with the In-Situ Adaptive Tabulation Algorithm (ISAT) [8] for the reduction of calculation times.

The condensation of ash vapours is modelled based on a mass transfer approach using the analogy of mass transfer to the convective heat transfer to determine the mass transfer coefficient  $\beta$  ([2]). Here,  $\dot{N}_{cond}$  represents the condensation flux per area,  $\alpha$  the heat transfer coefficient,  $c_p$  the heat capacity and  $\rho$  the density of the flue gas.

$$\dot{N}_{cond} = \frac{\alpha}{c_p \cdot \rho} Le^{m-1} \cdot (c_\infty - c_w) \quad (0.1)$$

The Lewis number  $Le$  as well as the exponent  $m$  were set to unity. While for  $c_\infty$  thermodynamic equilibrium calculations are carried out only under consideration of gaseous components, for the wall concentration  $c_w$  also liquid and solid phases at the surfaces have to be considered. In this work, the components  $KCl$ ,  $(KCl)_2$ ,  $K_2SO_4$ ,  $K_2CO_3$ ,  $NaCl$ ,  $(NaCl)_2$  and  $Na_2SO_4$  are considered as ash forming vapours.

Similarly to condensation, the calculation procedure is implemented to calculate the super-saturation of ash components in the flue gas to describe the nucleation of the ash vapour components as initial aerosol formation process. The nucleation rate is calculated according to Friedlander [5], which is based on the classical theory of nucleation:

$$I_{nuct} = 2 \cdot \frac{p_i}{\sqrt{2 \cdot \pi \cdot m_m \cdot k_B \cdot T}} \cdot V_m^{2/3} \cdot n_m \cdot \sqrt{\frac{\sigma \cdot V_m^{2/3}}{k_B \cdot T}} \cdot \exp\left(-\frac{16 \cdot \pi \cdot \sigma^3 \cdot V_m^2}{3 \cdot (k_B \cdot T)^3 \cdot \ln(S)^2}\right) \quad (0.2)$$

Here,  $p_i$  is the vapour pressure of the nucleating compound  $i$ ,  $m_m$  is the mass of a molecule calculated from the molar weight and Avogadro's number,  $k_B$  the Stefan–Boltzmann constant,  $V_m$  is the volume of a molecule calculated from the liquid density and Avogadro's number,  $n_m$  is the molecule concentration in the gas,  $\sigma$  is the surface tension of the nucleating compound,  $S$  is the saturation ratio ( $=p_i/p_{sat}$  with  $p_{sat}$  = saturation pressure of the nucleating compound).

Besides the formation of new particles by nucleation, the model also considers condensation of ash vapours on the surface of existing aerosol particles. For this purpose, the condensation flux is determined by

$$I_{cond} = \frac{F_i \cdot MW_i \cdot 6}{\rho_i \cdot d_p^3 \cdot \pi}$$

$$F_i = \frac{2 \cdot \pi \cdot D_i \cdot d_p \cdot (p_i - p_{i,sat}^*)}{k_B \cdot T} \quad (0.3)$$

with  $\rho_i$  as density of the condensing compound,  $D_i$  as diffusion coefficient of the condensing compound in the gas phase and  $MW_i$  as molecular weight of the condensing compound.

At the present state of development, a constant particle size ( $10^{-7}$  m) was assumed. Consequently, nucleation and condensation results in a change of the number of fine particles formed. The deposition of fine particles is considered by Fick's law for diffusion and thermophoresis is according to:

$$v_{th} = \frac{v_{gas} \cdot K_{th}}{T_p} \nabla T_{gas} \quad (0.4)$$

$K_{th}$  is the so-called thermophoretic coefficient, which is determined by Talbot [12],  $v_{gas}$  the viscosity of the gas,  $T_p$  the temperature of the particle and  $\nabla T_{gas}$  the local temperature gradient prevailing in the gas phase. Deposition mechanisms of fine particles due to inertial impaction, gravitational settling as well as diffusio-phoresis are neglected up to now, but it is known, that they only play a minor role [3].

#### DEPOSITION OF COARSE FLY ASH PARTICLES

The transport of coarse fly ash is modelled using an Lagrangian approach, which is already implemented in the CFD code FLUENT (Discrete Phase Model). Once a particle impacts the wall, either its kinetic energy is high enough to rebound from the surface, or the particle sticks to the wall. The total sticking probability  $p_{tot}$  of the particle depends on the properties (sticking probabilities  $p$ ) of the wall and the particle (Walsh [14])

$$p_{tot} = p_{wall} \cdot p_{part} + p_{wall} \cdot (1 - p_{part}) + (1 - p_{wall}) \cdot p_{part} \quad (0.5)$$

In this model, a viscosity approach for silica-rich particles and a melting approach for salt-rich particles is used. The sticking probability of silica-rich particles is calculated by the ratio of the reference viscosity  $\mu_{ref}$  and the viscosity of the particle at a certain temperature  $\mu_{part}$ . While the viscosity of the silica-rich material is determined using a viscosity model of Urbain [13], the reference viscosity  $\mu_{ref}$  is calculated as a function of the kinetic energy of the particle ([9]):

$$\log_{10}(\mu_{ref}) = -0.9699 \cdot \log_{10}\left(\frac{E_{kin}}{E_{kin,0}}\right) + 5.5962 \quad (0.6)$$

Here, the kinetic energies of the particles  $E_{kin}$  are normalized with the kinetic energy of the smallest particle ( $E_{kin,0} = 4.35e^{-11}$  J) studied in the work of Srinivasachar et al. [10].

The melt approach according to Backman [1] was applied for salt-rich particles and deposit layers. Based on comparisons of experimental data from biomass fired boilers with thermodynamic calculations, the author derived a criteria of 15% molten phase of the ash, below which the particles are not sticky (sticking probability  $p_{part, salts} = 0$ ). Above this melt fraction, a linear correlation of the stickiness from 0 to 100% is assumed up to a melt fraction of 0.7 (flow condition for the ash [1]). Within the CFD code, the amount of the melt fraction is calculated by thermodynamic equilibrium calculations considering condensed phases (salt mixtures, slags).

Since a particle and a deposit layer consist of a mixture of salt and silica-rich material, the melt approach as well as the viscosity approach were combined using the mass weighted average of the stickiness probability of both models.

If the coarse fly ash particles do not stick on the walls, they rebound. Depending on the particle velocity and the impact angle, this process causes erosion of the deposit layer. In order to consider these erosion effects by coarse fly ash particles, the model from Neilson and Gilchrist which considers the two known main mechanisms, brittle erosion and ductile erosion was implemented.

### 3. TEST RUNS

#### PLANT DESCRIPTION AND OPERATING CONDITIONS

In order to evaluate the deposit formation model described in chapter 2, test runs with two different plants, a 70 kW wood pellet fixed bed furnace and a 10 MW<sub>th</sub> biomass grate furnace including fuel analyses as well measurements of boiler heat load, air mass fluxes, conventional flue gas and dust measurements have been performed (**Fig.1**). In addition, in the case of the small-scale furnace the mass and chemical composition of grate ash, fly ash and hard ash in different plant zones have been measured and in the case of the medium-scale combustion plant deposit probe measurements have been carried out.

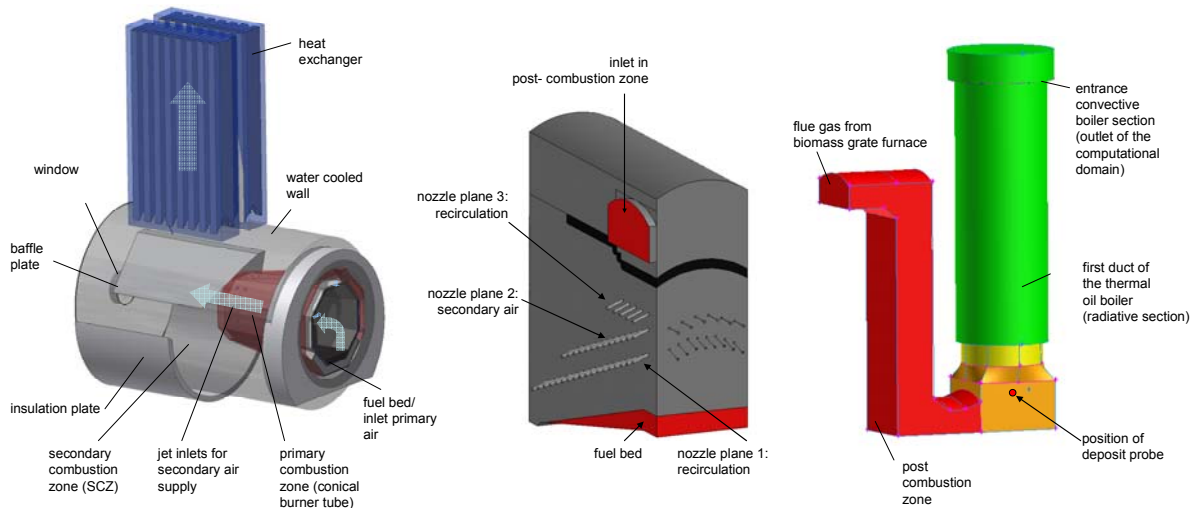


Fig. 1: Scheme of: left) the small-scale plant (70 kW boiler), middle) the 10 MW biomass grate furnace until the entrance in the post-combustion zone; right) the simulated section in the surrounding area of the deposit probe

Since the deposit formation simulations are very time-consuming, they have only been considered in the surrounding area of the deposit probe in the case of the medium-scale combustion plant. Within the frame of a previous simulation run, the kinetic conversion of sulphate precursors as well as the wall condensation fluxes in the biomass furnace were determined to gain the boundary conditions for the subsequent deposit formation simulation. While

Fig. 1 (middle) provides an overview about the biomass grate furnace,

Fig. 1 (right) illustrates the sections of the plant considered for the deposit formation simulations. The computational domain comprises the post combustion chamber of the furnace until the end of the radiation part of the thermal oil boiler. The deposit-probe measurements were carried in the entrance to the first duct of the thermal oil boiler (radiation part).

The most relevant operating conditions during the test runs as well as the chemical composition of the ash forming elements of the fuels are shown in Table 1. The small-scale plant

Fig. 1 (left) is divided into a primary combustion zone (PCZ) located in the rear part of the burner tube, which is cooled by the secondary air and a water cooled secondary combustion zone (SCZ). After the combustion of the fuel in the burner tube (PCZ), the released flue gas is mixed with secondary air supplied by nozzles at the exit of the burner tube. At the top of the SCZ, the heat exchanger tube bundle is installed. The automatic cleaning system for the heat exchanger tubes was not installed for the purpose of the model check.

Table 1: Operating conditions (left) and concentration of ash forming elements [mg/kg d.b.] in the fuels (right)

| parameter                      | units     | small-scale plant |         |             | medium-scale plant |       |       | element | wood pellets | fibre board | sawdust |
|--------------------------------|-----------|-------------------|---------|-------------|--------------------|-------|-------|---------|--------------|-------------|---------|
| fuel                           |           | wood pellets      | sawdust | fibre board |                    |       |       |         |              |             |         |
| net calorific value            | MJ/kg w.b | 17,00             | 12,45   | 16,92       | Si                 | 199   | 225   | 787     |              |             |         |
| fuel flow rate                 | kg/h      | 16                | 3.469   | 2.647       | Ca                 | 1.000 | 898   | 2.080   |              |             |         |
| C/H ratio                      | -         | 8,79              | 8,32    | 7,88        | Mg                 | 118   | 117   | 266     |              |             |         |
| ash content                    | wt.% d.b. | 0,40              | 0,31    | 1,49        | K                  | 402   | 318   | 639     |              |             |         |
| moisture content               | wt.% d.b. | 6,8               | 30,8    | 6,3         | Na                 | 38,2  | 13,3  | 239     |              |             |         |
| boiler load                    | kW        | 74                | 13.200  | 12.400      | S                  | 64,7  | 56,3  | 477     |              |             |         |
| primary air mass flux          | kg/h      | 31                | 6.903   | 7.630       | Cl                 | 44    | 19,2  | 190     |              |             |         |
| secondary air mass flux        | kg/h      | 122               | 18.502  | 19.620      | Zn                 | 9,88  | 8,51  | 35,1    |              |             |         |
| recirculation ratio            | -         | 0,00              | 0,34    | 0,40        | Pb                 | 5     | 0     | 0       |              |             |         |
| total air ratio                | -         | 1,79              | 1,84    | 1,80        | Fe                 | 26,3  | 7,44  | 150     |              |             |         |
| primary air ratio              | -         | 0,36              | 0,50    | 0,50        | Mn                 | 168   | 83,8  | 81,6    |              |             |         |
| adiabatic flue gas temperature | °C        | 1.329             | 898     | 861         | Al                 | 34,2  | 11    | 846     |              |             |         |
| temperature cooling medium     | °C        | 62                | 300     | 300         | P                  | 49,5  | 19,2  | 0       |              |             |         |
|                                |           |                   |         |             | Cu                 | 0     | 0,697 | 2,05    |              |             |         |
|                                |           |                   |         |             | Ti                 | 1,89  | 0     | 0       |              |             |         |

## TEST RUN RESULTS

Table 2 (left) shows the total mass of hard deposits, fine particle emissions and coarse fly ash emissions of the small-scale furnace over an operation time of 69 h at full load. The overall mass balance regarding ashes found in comparison to the ash input with the fuel closes well which underlines the plausibility of the measurements performed.

Table 2: Mass of ash deposits and emissions of the small- (left) and the medium scale furnace (right)

| measurements in the small scale plant over an operation time of 69h       |          | measurements in the medium-scale plant over a period of 9 h                |             |         |
|---|----------|--|-------------|---------|
| position  | mass [g] | mass fluxes [kg/h]   |             |         |
| bottom ash inside the burner tube   | 420      |  |             |         |
| bottom ash on the insulation plate  | 1698     |  |             |         |
| deposits on burner tube walls   | 52       |  |             |         |
| deposits on the baffle plate  | 131      |  |             |         |
| deposits on cooled boiler walls of the SCZ                                | 10       |  |             |         |
| ash on window   | 33       |  |             |         |
| ash on the surface of the heat exchanger                                  | 268      |  |             |         |
| ash in the exhaust gas receiver   | 207      |  |             |         |
| coarse fly ash emissions  | 194      |  |             |         |
| fine particle emissions   | 110      |  |             |         |
| sum of ash samples  | 3124     |  |             |         |
| ash input according to fuel analysis                                      | 3165     |  |             |         |
| rate of recovery [%]  | 99       |  |             |         |
| flue gas measurements of boiler outlet (emission to atmosphere)           |          | position   | fibre board | sawdust |
| CO [ppm]  | 26       | bottom ash on the grate  | 1,43        | 16,75   |
| excess oxygen in flue gas [Vol.-%, dry base]                              | 9,3      | coarse fly ash emissions   | 3,39        | 17,58   |
| flue gas temperature of boiler exit [°C]                                  | 266      | fine particle emissions  | 1,62        | 2,64    |
| total dust [mg/Nm <sup>3</sup> ] (13 Vol.-%O <sub>2</sub> , dry flue gas) | 21,2     | ash input according to fuel analysis                                       | 6,44        | 36,97   |
| PM1 [mg/Nm <sup>3</sup> ] (13 Vol.-%O <sub>2</sub> , dry flue gas)        | 7,65     | flue gas measurements of economiser outlet (before ESP)                    |             |         |
|   |          | CO [ppm]   | 8,9         | 10,1    |
|   |          | excess oxygen in flue gas [Vol.-%, dry base]                               | 9,5         | 9,7     |
|   |          | flue gas temperature of boiler exit [°C]                                   | 185,5       | 184,3   |
|   |          | total dust [mg/Nm <sup>3</sup> ] (13 Vol.-%O <sub>2</sub> , dry flue gas)* | 347         | 115,8   |
|   |          | PM1 [mg/Nm <sup>3</sup> ] (13 Vol.-%O <sub>2</sub> , dry flue gas)*        | 52,13       | 37,5    |

\* measurement at the outlet of the economiser of the plant

Furthermore, the measurements of the medium-scale plant are shown. Based on total dust and fine particle measurements, the amount of bottom ash was determined from the ash input according to the fuel analysis.

In all cases, the burnout conditions were very good as indicated by the low CO emissions.

The release of the ash vapours and the entrainment of coarse fly ash particles from the fuel bed depend on the chemical composition of the fuel and the local conditions in the fuel bed (local temperature, stoichiometry and flue gas velocity in the bed). If experimental data regarding mass and elemental balances of the biomass combustion plants are available in a good quality, the release of ash vapours and fly ash particles can be determined from test runs. Since this precondition for the mass balance data was fulfilled for the small-scale combustion plant, the released mass of the aerosol forming elements K, Na, Cl and S from the fuel bed was calculated from the difference of the mass of these elements found in the bottom ash fractions and in the coarse fly ash fractions (under the assumption that the coarse fly ash has the same composition than the bottom ash when entrained from the fuel bed) and the amount of these elements supplied with the fuel. The amount of coarse fly ash entrained from the fuel bed was calculated by subtracting the amount of bottom ash formed and the amount of ash vapours released from the total mass of ash supplied with the fuel. Since the deposits from the exhaust gas channel mainly consist of coarse fly ash particles, their chemical composition was used to calculate the particle stickiness properties of the model.

In the case of the medium-scale combustion plant, the release of the ash vapours had to be estimated from previous lab-scale investigations using similar biomass fuels and operational conditions [7], since the elemental ash balance could not be closed. Table 3 right gives an overview about the percentage of aerosol forming elements which are released from the fuel to the gas phase. Additionally, the chemical composition of the coarse fly ash particles entrained from the fuel bed is shown, which is used to calculate the coarse fly ash stickiness of the model. In the case of the small-scale furnace the chemical composition of coarse fly ash particles found in deposits from the exhaust gas channel has been taken, in the case of the medium-scale plant the coarse fly ash particles found in the cyclone after the economiser were used.

Table 3: Percentage of aerosol forming elements (wt.%) released from the fuel (right) and chemical composition of the coarse fly ash particles (wt.% d.b.) entrained from the fuel bed (left)

| element | small-scale plant |             |         | medium scale plant |             |         |
|---------|-------------------|-------------|---------|--------------------|-------------|---------|
|         | wood pellets      | fibre board | sawdust | wood pellets       | fibre board | sawdust |
| Si      | 3.35%             | 4.92%       | 4.74%   |                    |             |         |
| Ca      | 25.20%            | 21.60%      | 23.80%  |                    |             |         |
| Mg      | 3.39%             | 2.71%       | 2.87%   |                    |             |         |
| K       | 12.30%            | 4.13%       | 4.43%   |                    |             |         |
| Na      | 0.65%             | 0.61%       | 0.48%   |                    |             |         |
| S       | 0.34%             | 1.35%       | 1.41%   |                    |             |         |
| Cl      | 0.07%             | 0.45%       | 0.48%   |                    |             |         |
| Zn      | 0.04%             | 0.14%       | 0.14%   |                    |             |         |
| Pb      | 0.00%             | 0.02%       | 0.02%   |                    |             |         |
| Fe      | 0.65%             | 1.18%       | 1.31%   |                    |             |         |
| Mn      | 4.78%             | 1.75%       | 2.00%   |                    |             |         |
| Al      | 0.54%             | 4.03%       | 3.40%   |                    |             |         |
| P       | 1.26%             | 0.93%       | 0.94%   |                    |             |         |
| Ti      | 0.04%             | 0.00%       | 0.00%   |                    |             |         |

| element | small-scale plant |             |         | medium scale plant |             |         |
|---------|-------------------|-------------|---------|--------------------|-------------|---------|
|         | wood pellets      | fibre board | sawdust | wood pellets       | fibre board | sawdust |
| K       |                   | 25          | 37      | 45                 |             |         |
| Na      |                   | 58          | 75      | 35                 |             |         |
| S       |                   | 87          | 85      | 85                 |             |         |
| Cl      |                   | 96          | 100     | 100                |             |         |

#### 4. SIMULATION RESULTS

In order to illustrate the chemical interactions of the ash forming vapours in the plants, some fundamental simulation results regarding the flue gas temperature and the kinetically limited conversion of sulphate precursors ( $H_2S$ , S,  $SO_2$ ) to  $SO_3$  and sulphates is shown. Afterwards, selected results regarding condensation, fine particle formation as well as the deposition behaviour of coarse fly ash particles are given. Finally, in the case of the small-scale plant, simulations have been performed considering all sub-models implemented in the deposit formation model. Ash deposit rates as well as the emissions of coarse fly ash and fine particles are compared with measurements to check and evaluate the simulations.

##### FLUE GAS TEMPERATURE AND SULPHUR KINETICS

**Fig. 2** shows the molar ratio of gaseous sulphates and  $SO_3$  in the flue gas in a vertical cross section of the plant, which is formed by the oxidation of sulphate precursors (S,  $H_2S$ ,  $SO_2$ ). Despite of the high gas temperatures inside the burner tube (see **Fig. 2 a**) the lack of oxygen in the PCZ inhibits the oxidation of the released volatile sulphur species. Subsequently, after mixing with secondary air this

restriction is removed and the concentration of sulphates and  $\text{SO}_3$  increases. Subsequently, due to the formation of fine particles and condensation of the ash vapours on the walls, the molar fraction decreases.

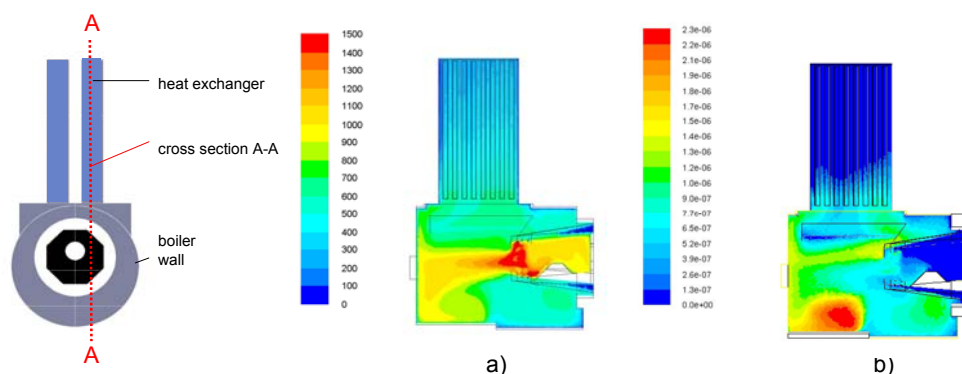


Fig. 2: a) Calculated flue gas temperatures ( $^{\circ}\text{C}$ ) in the vertical cross section A-A of the boiler; b) molar fraction [mol/mol] of gaseous sulphates and  $\text{SO}_3$  formed by the oxidation of sulphur precursors

In **Fig. 3** the conversion rates of the released sulphate precursors to sulphates and  $\text{SO}_3$  are shown for all test cases. Here, the molar ratio is defined as the sum of sulphur located in fine particles and deposits to the amount of sulphur in the released sulphur precursors. It can be seen, that the sulphur conversion varies between 40 % and 87 %.

| plant                | fuel        | sulphur conversion [mol/mol] |
|----------------------|-------------|------------------------------|
| small-scale furnace  | wood pellet | 0.66                         |
| medium-scale furnace | sawdust     | 0.87                         |
| medium-scale furnace | fibre board | 0.40                         |

Fig. 3: Conversion rates of sulphur precursors to gaseous sulphates and  $\text{SO}_3$

#### FINE PARTICLE FORMATION AND CONDENSATION

Due to the oxidation of  $\text{SO}_2$  gaseous sulphates are formed in the SCZ (**Fig. 4, left**).

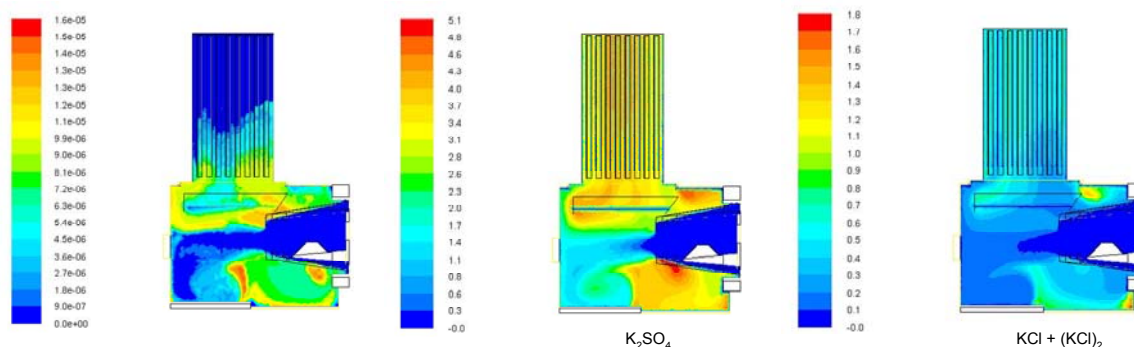


Fig. 4: Simulation results regarding left) the concentration field of gaseous  $\text{K}_2\text{SO}_4$  [ $\text{kmol}/\text{m}^3$ ] middle) the fine particles formed by nucleation/condensation of gaseous  $\text{K}_2\text{SO}_4$  [ $\text{mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ], left) the fine particles formed by nucleation/condensation of  $\text{KCl}$  and  $(\text{KCl})_2$  [ $\text{mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ]

Simultaneously, if the temperature is below  $900^{\circ}\text{C}$ , the amount of gaseous  $\text{K}_2\text{SO}_4$  is decreased by the formation of fine particles. Hence, the maximum sulphate concentrations can be found in a boundary zone between the formation zone of sulphates and regions of fine particle formation. In **Fig. 4 (middle)**, the concentration field of the fine particles formed by nucleation and condensation of  $\text{K}_2\text{SO}_4$  as well as the concentration field of fine particles formed by  $\text{KCl}$  and the dimer  $(\text{KCl})_2$  (**right**) are shown. The different fine particle formation behaviour which can be observed in the burner tube and in the SCZ is based on the different flue gas temperatures in these regions. Since the temperatures inside the burner (see **Fig. 2 a**) are above  $1,000^{\circ}\text{C}$ , the initial formation of fine particles can only occur at the outlet of the burner. In the cooled SCZ, sulphates are formed in the hot zone near the outlet of the PCZ. Fine particles consisting of chlorides are formed in the boundary zones near the cooled wall.

In **Fig. 5** the total concentration field of the fine particles is illustrated. Furthermore, the chemical compositions of the fine particles are compared with fine particle measurements. It is found that the amount of sulphate in the fine particles is slightly overpredicted by the model. The averaged concentration of the sum of all fine particles at the furnace outlet ( $9.92 \text{ mg/Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ) is in good agreement with experimental data ( $7.65 \text{ mg/Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ).

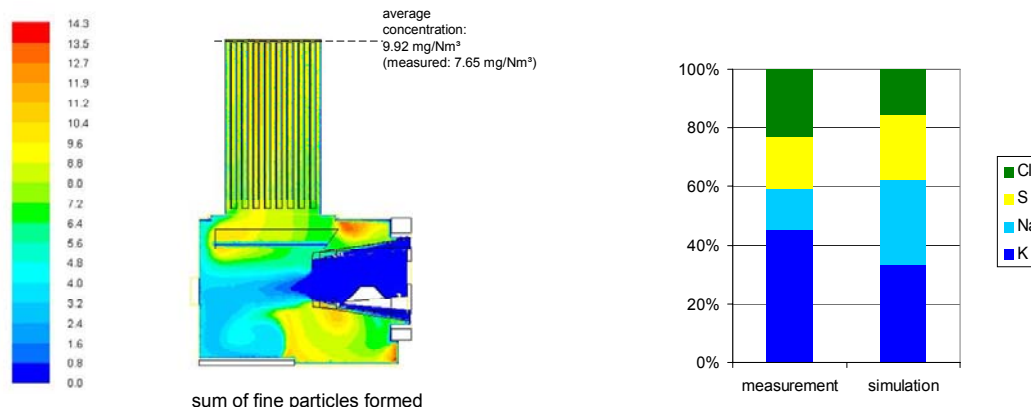


Fig. 5: Simulation results regarding aerosol formation: left) total fine particle emissions [ $\text{mg/Nm}^3$  dry flue gas, 13 %  $\text{O}_2$  ]; right) comparison of measurement and simulation results regarding the chemical composition of fine particles [mol-% d.b.]

**Fig. 6 (left)** illustrates the wall temperatures of the simulated sections of the medium-scale plant (furnace section and radiative boiler section), the overall deposit mass flux caused by condensation and fine particle precipitation (middle) as well as the mass flux by KCl precipitation of fine particles and condensation on the walls (right).

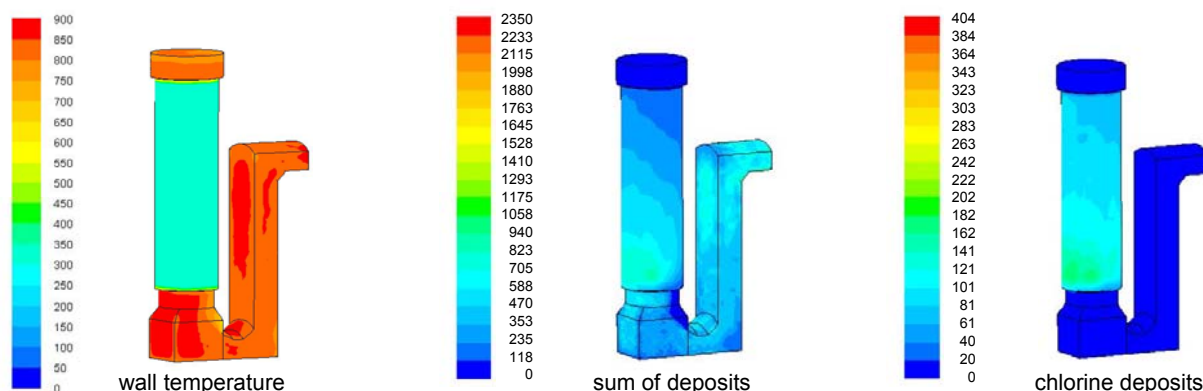


Fig. 6: left) wall temperature [ $^{\circ}\text{C}$ ] of the simulated section of the medium-scale combustion plant; middle) overall deposit mass flux caused by condensation and fine particle precipitation [ $\text{mg/m}^2\text{h}$ ]; right) deposit mass flux of chlorine [ $\text{mg/m}^2\text{h}$ ] caused by condensation and fine particle precipitation; fuel: saw dust

A different deposit formation behaviour can be observed at the cooled walls the thermal oil boiler and the insulated walls of the post-combustion chamber. Due to the different wall temperatures and the different vapour pressures of the chlorines and sulphates at the surfaces of the walls the deposition of chlorides occurs only at temperatures below  $600^{\circ}\text{C}$ . In contrast, sulphate containing fine particles are formed already at temperatures below  $900^{\circ}\text{C}$ . Within this boiler section, the primary deposit formation mechanism caused by ash forming vapours is condensation, thermophoresis and diffusion of fine particles play only a minor role.

In **Fig. 7** a comparison of experimental and numerical data regarding the chemical composition of the deposit probe concerning the considered elements K, Cl, Na and S is shown. The deposits were formed by precipitation of fine particles as well as by condensation of ash forming vapours on the cooled wall. While for sawdust, a good agreement can be found, for the fuel fibre board, the amount of sulphates calculated is somewhat to low.



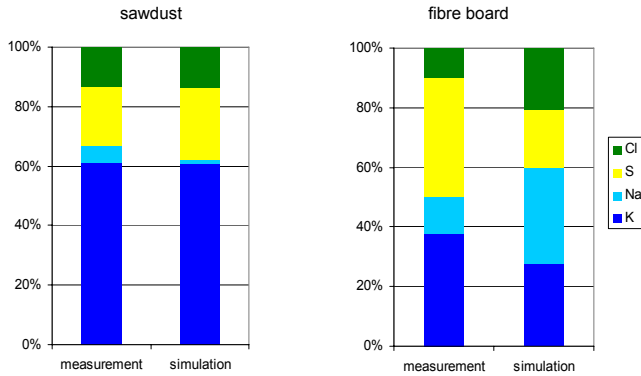


Fig. 7: Comparison of experimental and numerical results regarding the chemical composition [mol -%] of the deposit layer formed; left: sawdust, right: fibre board

#### DEPOSITION OF COARSE FLY ASH PARTICLES

**Fig. 8 a** shows calculated deposition mass fluxes caused by the impaction of coarse fly ash particles in the small-scale plant (other deposition processes are not considered). The main part of the particles is captured by the inner hot walls of the burner tube. Also in the SCZ the model predicts a high sticking probability of the particles impacting on the insulation plate. Only 3 wt.% of the released particles reach the outlet of the heat exchanger (measured value approximately 50 %). As a consequence, the influence of the erosion model can be neglected within this simulation because the high stickiness results in no relevant particle rebound rates after their impact on the walls.

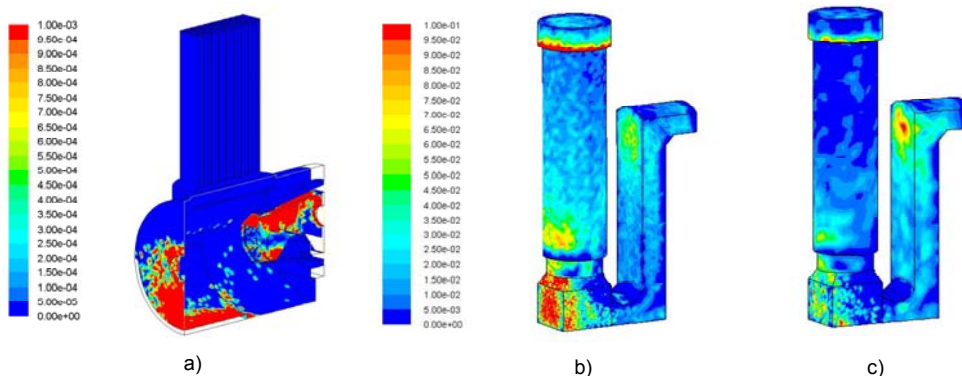


Fig. 8: Deposit mass flux of coarse fly ash particles ( $\text{kg}/\text{m}^2\text{h}$ ); a) small-scale plant (wood pellets), b) medium-scale plant (fibre board); c) medium-scale plant (sawdust)

Since the mass fraction of silica-rich components of the particles amounts to 98 %, the stickiness of the particles is almost entirely calculated by the viscosity approach. Consequently, the model predictions regarding the reference viscosity should be improved. For this purpose, experimental investigations concerning the reference viscosity of the stickiness approach regarding silica-rich particles (e.g. described in [9]) are necessary to improve the stickiness approach used.

**Fig 8 b)** and **c)** show the deposit build-up caused by the impaction of coarse fly ash particles in the simulated sections of the medium-scale plant. Also in this case, the stickiness of the coarse fly ash particles is over-predicted, since most of the particles are captured on the walls of the post-combustion channel. As a consequence, measured coarse fly ash emissions after the economiser (sawdust:  $78 \text{ mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ; fibre board:  $294 \text{ mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ) are significant higher than the numerical results at the end of the computational domain (sawdust:  $33 \text{ mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ; fibre board:  $176 \text{ mg}/\text{Nm}^3$  dry flue gas, 13 %  $\text{O}_2$ ).

#### OVERALL DEPOSIT BUILD-UP

In **Fig. 9** the experimental data from test runs of the small-scale furnace are compared with a simulation considering condensation of ash vapours, aerosol formation and deposition as well as deposition of coarse fly ash particles (overall model).

|                                      | measured deposit<br>mass<br>[g] | simulation result<br>(overall model)<br>[g] |
|--------------------------------------|---------------------------------|---|
| deposition flux                      | 535                             | 910   |
| emission of fine particles (PM1)     | 110                             | 98  |
| emission of coarse fly ash particles | 194                             | 25  |

Fig. 9: Comparison of experimental and numerical data (from the overall model) regarding deposition and emission results

The comparison shows that the measured deposition flux is lower than the simulation result almost by a factor of two, while the predicted fine particle emissions are in good agreement with the measurements. As already discussed in the previous section, the reasons can mainly be found in the overestimation of the stickiness of the coarse fly ash particles. In addition, effects that have not been taken into account by the model contribute to deviations to the numerical results, e.g. particle re-entrainment by shear forces of the flue gas or shedding of the deposit layers.

Since the accuracy of the model regarding the stickiness of the coarse fly ash particles is not sufficient to describe the chemical composition and thickness of deposit layers, its influence on heat transfer in the furnace and the boiler is not discussed in this paper. During the test run of the small-scale furnace, an increase of the flue gas temperature of 26 K was measured over a period of 40 h. It follows that the heat flux to the boiler is reduced by 1 kW after this time (1.4 % of the boiler output, 25 W/h).

## 5. SUMMARY AND CONCLUSIONS

The numerical results gained by the application of the deposit formation model for two different plant sizes of fixed bed biomass boilers and three different fuels illustrate the contribution of different mechanisms to deposit build-up and aerosol formation. The influence of chemical reactions, aerosol formation, condensation as well as the deposition of coarse fly ash particles is shown.

Comparisons of model predictions with experimental findings regarding the emissions of coarse fly ash particles show, that the stickiness of the silica-rich particles is over-predicted by the model. The assumption of a material independent reference-viscosity, which was developed for coal ashes, seems not to be applicable for biomass ashes. In order to increase the accuracy of the stickiness model for silica rich particles, the experimental determination of the reference viscosity as a function of the kinetic energy of the particles for different biomass fuels is necessary.

The predictions of the aerosol formation model in combination with the condensation model regarding fine particle concentrations and chemical compositions are in good agreement with experiments. The chemical composition of the deposit probe measurements of the medium-scale plant are also in good agreement with simulations in case of sawdust. In the case of the fibre board residues, the sulphate concentration is under-predicted, which may be due to the sulphur kinetics used. Nevertheless, some deviations from the measurement results must be expected, since the model contains some simplifications (e.g. constant material properties) as well as no modelling of the particle size distribution. Furthermore, temperature fluctuations in the flue gas prevailing during the test runs have an impact on the ash chemistry (e.g. sulphation rates).

Concluding, it can be stated that an advanced tool for an efficient design of biomass furnaces and boilers is under development. The influence of the fuel fired and operation conditions on deposit and aerosol formation processes can be investigated in detail considering the relevant mechanisms already in the design phase of the plant and thus allowing for appropriate measures in order to reduce ash deposit formation and to take proper choices regarding material selection in terms of corrosion issues. The model can already deliver quantitatively correct results regarding fine particle formation and condensation. In order to achieve better predictions regarding the overall deposit build-up, a better description of the stickiness behaviour of coarse fly ash particles should be implemented. As further model development steps, the consideration of aerosol particle size distributions as well as the deposit build-up in heat exchanger tube bundles are foreseen. Moreover, experiments in a specially designed biomass-fired drop-tube furnace with an implemented deposit probe are planned to achieve reliable data regarding erosion and stickiness.

## 6. ACKNOWLEDGEMENTS

The work presented was supported by the Kplus program of the BIOENERGY 2020+. The financial support of the Austrian Research Promotion Agency as well as of the state governments of Lower Austria and Styria and of the City of Graz as well as of the companies BIOS BIOENERGIESYSTEME GmbH, MAXXTEC AG, Hovalwerk AG, KWB Kraft und Wärme aus Biomasse GesmbH is gratefully acknowledged. The deposit measurements were performed in cooperation with Hovalwerk AG, Liechtenstein, BIOS BIOENERGIESYSTEME GmbH, Austria and MAXXTEC AG, Germany.

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